The University of Michigan College of Engineering Curriculum Committee

Agenda
January 09, 2007
1:30-3:00 p.m.
GM Room
Fourth Floor Lurie Engineering Center

- 1. Approval of Minutes from December 12, 2006 Meeting
- 2. The Diversity and Outreach Committee Presentation
- 3. Course Approvals

University of Michigan College of Engineering Curriculum Committee Meeting Tuesday December 12, 2006 1:30-3:00 p.m. Lurie Engineering Center GM Room Minutes

Richard Robertson called the meeting to order at 1:40 p.m.

Members Present: R. Robertson, J. Boyd, G. Hulbert, D. Karr, K. Kearfott, C. Lastoskie, Y. Liu, P. Mazumder, H. Peng, R. Rogers, M. Solomon, R. Sulewski T. Teorey, A. Yagle

Members Absent: L. Bernal, A. Hunt, G. Herrin, S. Pang, K.Patel

Guests: Ted Spencer (Associate Vice Provost and Executive Director, Undergrad Admissions), Chris Lucier (Director of Recruitment and Operations), Feodies Shipp (Assistant Director)

Motion to approve the minutes of the last meeting

The minutes of the last meeting were approved

<u>Creation of Subject area for the Financial Engineering Degree Program – Huei Peng</u>

Information regarding this was included in the meeting packet.

Huei Peng spoke requesting a new course, Fin Eng 591, emphasizing that this is only a new course and not a new program.

"As a result of its interdisciplinary nature, the College of Engineering Financial Engineering program is actively engaged in expanding its curricular activities. This includes the development of various study templates, the development of new courses, such as our new, mandatory summer program, and the cross-listing of a variety of courses with the Ross Business School and the Math and Statistics Departments. To be able to engage in these activities expeditiously, the Financial Engineering program needs to have its own subject area."

Dick Robertson asked for a motion to approve this course. Moved and Seconded. This course, Fin Eng 591, was approved.

Course Approval Forms

Richard Robertson called for a motion to approve the following courses. This was moved and seconded.

These Courses Were Approved

AOSS 101 (X-Listed with ASTRO 103) New Course

AOSS 480 New Course

FINENG 591 New Course

Undergraduate Admissions Overview – Christopher Lucier

Information Packets were handed out regarding this information. Christopher Lucier, Ted Spencer and Feodies Shipp, from the Office of Undergraduate Admissions, presented this information geared toward the College of Engineering. Members of this Committee were invited to ask questions.

<u>Adjournment:</u> Motion to adjourn was made and seconded <u>Motion carried (approved)</u>

Next Meeting January 9, 2007 GM Room – Fourth Floor LEC

COURSE APPROVAL FORMS

For January 09, 2007 CoE CC Meeting

ME 320	Modification – Changing Prerequisites from: ME 235 and ME 240 to: MATH 215, ME 235 and ME 240.
ME 450	Modification – Changing Credit Restrictions from: ME 495 is not to be elected concurrently. Not open to graduate students. to: May not be taken concurrently with ME 495. Not open to graduate students.
ME 495	Modification – Changing Prerequisites from: ME 360, ME 395; preceded or accompanied by ME 350 to: ME 360, ME 395, P/A ME 335 and ME 350; Changing Credit Restrictions from: Recommend ME495 not be elected concurrently. Not open to graduate students to: May not elect ME 450 concurrently. Not open to graduate students.
ME 539	Modification – Changing Title from: Heat Transfer in Porous Media to: Heat Transfer in Physics; Changing Description; Changing Prerequisites from: ME 335 or equivalent to: ME 235, 335.
ME 574	New Course
ME 580	New Course

THE UNIVERSITY OF MICHIGAN -- COLLEGE OF ENGINEERING Course Approval Request

College Curriculum Committee, 1420 Lurie Engineering Center Building

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Form	Number
1786	

Action Requested

New CourseModification of Existing CourseDeletion of Course

Complete the following sections:

New Courses - B & C completely

Modifications - A modified information, B & C completely

Deletions - A & C completely

Date 11/9/2006 Effective Winter 2007

	A. CURRENT LISTING	B. REQUESTED LISTING
	Home Department Div # Course Number	Home Department Div # Course Number
П		Mechanical Engineering 280 320
	Cross Listed Course Information	Cross Listed Course Information
П	Course Title	Course Title
_		Fluid Mechanics I
	TITLE	TITLE ABBRE- VIATION Transcript Transcript Transcript Fluid Mechanics I Fluid Mechanics I
_	Max = 20 Spaces	Max = 20 Spaces
	Course Description	Course Description for Official Publication (Max = 50 words) Control volume analysis; continuity, momentum, angular momentum, and energy equation. Dimensional analysis and similitude. Introduction to differential analysis; kinematics; fluid statics; inviscid flow; potential flow; simple viscous incompressible flow; lift and drag. Steady one-dimensional compressible flow.
	PROGRAM OUTCOMES: a b c d e f g h i j k	PROGRAM OUTCOMES: ⊠a □b ⊠c □d ⊠e ⊠f □g ⊠h ⊠i ⊠j ⊠k
	Degree Requirements O Degree Requirement O Free Elective O Tech Elective	Degree Requirements O Degree Requirement O Free Elective O Other O Tech Elective
Х	Prerequisites ME235 and ME240 © Enforced © Advised	Prerequisites MATH215, ME235, and ME240 © Enforced ® Advised
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C.	Repeatability (Indi Research, Dir. Study, Dissertation: Is this course repeatable? Yes No Maximum Hours? Maximum Times? Can it be repeated in the same term? Yes No	Printing Information Print the course in the Bulletin (Optional) Print the course in the Time Schedule
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	Approval	Submitted By: Me Home Dept. Cross-listed Dept.
	Curriculum Comm.	Name, Signature & Department
	Faculty Rackham Cross listed Unit 1 Cross listed Unit 2	Home Dept. Cross-listed Dept(s). Mechanical Engineering

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MATH 215 introduces vector algebra, vector functions, functions of seve ntegrals and their applications; Green's theorem and Stokes theorem. Th	nese are important concepts for students to have learned prior
to taking ME320.	
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Are any special resources or facilities required for this course?	☐ Yes ☒ No
Detail the Special requirements	
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ME320 COURSE PROFILE

DEGREE PROGRAM: Mechanical Engineering

COURSE NUMBER: ME320	COURSE TITLE: Fluid Mechanics I
REQUIRED COURSE OR ELECTIVE COURSE: Required	TERMS OFFERED: Fall, Winter
TEXTBOOK / REQUIRED MATERIAL: Fundamentals of Fluid Mechanics, B.R. Munson, D.F. Young, T.H. Okiishi, (4th edition) 2002, Wiley	PRE / CO-REQUISITES: ME235: Thermodynamics I; ME240: Introduction to Dynamics and Vibrations; MA215: Calculus III; MA216: Introduction to Differential Equations
COGNIZANT FACULTY: R. Akhavan	COURSE TOPICS:
BULLETIN DESCRIPTION: Fluid statics; conservation of mass, momentum, and energy in fixed and moving control volumes; steady and unsteady Bernoulli's equation; differential analysis of fluid flow; dimensional analysis and similitude; laminar and turbulent flow; boundary layers; lift and drag; introduction to commercial CFD packages; applications to mechanical, biological, environmental, and micro-fluidic systems.	 Fluid properties, fluid forces, and flow regimes. Fluid statics. Fluid statics. Flow kinematics. Conservation of mass, momentum and energy in fixed, deforming, and moving control volumes. The steady and unsteady Bernoulli equation along and normal to a streamline. Similitude, dimensional analysis, and modeling; important nondimensional groups in fluid mechanics. Conservation of mass and momentum expressed through differential analysis. Viscous flow in pipes and channels (laminar and turbulent flow regimes, the Moody chart, head-loss equation). External flow 9boundary layer concept, lift and drag, pressure and friction drag, streamlining and drag reduction). Introduction to commercial CFD package. Sample applications to mechanical biological, environmental, and micro-fluidic systems.
COURSE STRUCTURE/SCHEDULE: Lecture: 2 days per week at 1.5 hours	

COURSE OBJECTIVES: for each course objective, links to the Program Outcomes are identified in brackets.	 To teach basic fluid properties (density, viscosity, bulk modulus), flow forces (pressure, shear stress, surface tension), and flow regimes (laminar/turbulent, compressible/incompressible, steady/unsteady) [1, 5]. To teach how force is transmitted in static fluids [1, 5]. To teach conservation of mass, momentum, and energy in fixed, deforming, and moving control volumes [1, 5, 12]. To teach conservation of mass and momentum through differential analysis in simple geometries [1, 5, 12]. To teach conservation of mass and momentum through differential analysis in simple geometries [1, 5, 12]. To teach techniques of dimensional analysis, similitude, and modeling, and introduce the important non-dimensional groups in fluid mechanics [1, 2, 5, 9]. To teach application of the above concepts to internal and external flows, and introduce the boundary layer concept, lift and drag, flow separation, and drag reduction fundamentals [1, 3, 5, 9, 10]. To teach the use of commercial CFD packages [4, 6, 9, 10, 11]. To teach examples of applications of above concepts in mechanical, biological, environmental, and micro-fluidic systems [8, 9, 10, 11].
COURSE OUTCOMES: for each course outcome, links to the Course Objectives are identified in brackets.	 Ability to identify or predict the flow regime in a given engineering system based on consideration of the governing non-dimensional groups [1, 6, 8, 9]. Ability to calculate the hydrostatic forces and moments on planar and curved submerged and floating surfaces [1, 2, 8]. Ability to calculate the hydrostatic forces and moments on planar and curved submerged and floating surfaces [1, 2, 8]. Ability to construct an appropriate (fixed, deforming, or moving) control volume for a given engineering system and apply the principles of conservation of mass, momentum, and energy to this control volume [1, 2, 3]. Ability to decide when appropriate to use ideal flow concepts and the Bernoulli equation [1, 3, 4]. Ability to present data or governing equations in non-dimensional form, design experiments, and perform model studies [6, 1, 7]. Ability to solve for internal flow in pipes and channels through simple solutions of the Navier-Stokes equations, the Moody chart, the head-loss equation, or commercial CFD packages [5, 6, 7, 8]. Ability to solve for external flow, evaluate lift and drag, know when there is possibility of flow separation, apply streamlining concepts for drag reduction by using experimental correlations or commercial CFD packages [7, 8]. An understanding of how fluid mechanics applies to mechanical, biological, environmental, and micro-fluidic systems [9].
ASSESSMENT TOOLS: for each assessment tool, links to the course outcomes are identified	 Regular homework assignments Exams

PREPARED BY: R. Akhavan LAST UPDATED: November 2, 2006

THE UNIVERSITY OF MICHIGAN -- COLLEGE OF ENGINEERING **Course Approval Request**

College Curriculum Committee, 1420 Lurie Engineering Center Building

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Form	Number
1736	

Action Requested

New CourseModification of Existing CourseDeletion of Course

Complete the following sections:
New Courses - B & C completely
Modifications - A modified information, B & C completely
Deletions - A & C completely

Date 4/21/2006 Effective Winter 2007

A. CU	RRENT LISTING				B. RI	QUESTED LISTI	NG		
Home Depart	tment	397093334	Div#	Course Number	Home Dep	artment cal Engineering	warmen and the second	Div#	Course Number 450
Cross Listed C	Course Information				_	Course Information			
Course Title					Course Title	nd Manufacturing	III		
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VIATION	Transcript Max = 20 Spaces				VIATION	Transcript Max = 20 Spaces	Des/Mfg III		
Course Descr	7				A mecha exposed layout ar of study instruction	ription for Official Publication inical engineering to the design proceeding to the design proceeding to the design proceeding to the design proceeding the design mechanical faculty and incommend two labs.	design project b cess from conce are proposed for engineering and	pt through a rom the diff d reflect the	analysis to erent areas expertise of
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	ME350, ME360, and ME © Enforced © Advised	=395		(1990) Wall Co	Prerequisites	⊙ Enforced ⊝ Advise	ed		
Credit I Restrictions	ME495 is not to be elected	concurrently. Not ope	en to graduate st	udentss.	Credit Restrictions	May not be taken concurre	ently with ME495. Not o	pen to graduate	students.
Level of Cred Undergrad Rackham (Non-Rckhr	only Ugrad or N Grad All Credit t m Grad Rckhm Gi	Non-Rokhm Grad types rad w/add'l Work	Credit Hours Min Max	Contact Hrs/Wk Number of Wks	Level of Cro ☐ Undergrae ☐ Rackham ☐ Non-Rckh	d only Ugrad or N Grad All Credit m Grad Rckhm G	Non-Rokhm Grad lypes rad w/add'l Work	Credit Hours Min Max 4 4	Contact Hrs/Wk 8 Number of Wks 14
Is this course Maximum	eatability (Indi Research, I e repeatable? () Yes (: Hours? Maxir repeated in the same term	No No mum Times?	ii.		Printing I	nformation ⊠ Print the c (Optional) ⊠ Print the c	course in the Bulletin course in the Time Scheo	dule	
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The ME program does not recommend that students take ME450 and ME495 concurrently. For the past couple of years, this has been communicated to students on an individual basis when the come to the ASO to talk about their studies. For students who wish to take both at the same time, they are asked to see the UG Program Director to talk about their situation and why they need/desire to take this undesirable combination of courses. Typically, their course load is evaluated so that it, at most, includes 450, 495 and no more than one technical course.
At issue is that the 450/495 course combination not only adversely impacts the particular student taking these two courses, but even more importantly can adversely impact the group member of that student in both of the courses.
The proposed change would formalize the existing advising process and better help students develop their plan of study.
Are any special resources or facilities required for this course?
Detail the Special requirements
Unchanged since 1997-98

COURSE NUMBER: ME450	COURSE TITLE: Design and Manufacturing III
REQUIRED COURSE OR ELECTIVE COURSE: Required	TERMS OFFERED: Fall, Winter
TEXTBOOK / REQUIRED MATERIAL: N/A	PRE / CO-REQUISITES: ME 350: Design and Manufacturing II, ME 360: Modeling, Analysis and Control of Dynamic Systems, ME 395: Laboratory I. May not be taken concurrently with ME 495. Not open to graduate students.
COGNIZANT FACULTY: S. Skerlos	COURSE TOPICS:
which the student is exposed to the design process from concept through analysis to layout and report. Projects are proposed from the different areas of study within mechanical engineering and reflect the expertise of instructing faculty. Three hours of lecture and two laboratories.	 Generation of project specifications and solutions in a team environment. Systematic design procedures include: Definition of project requirements; Research of design problem background and state-of-the art; Development of project proposal with defined targets and scope of work; Development of quantitative design specifications from qualitative problem statement; Generation and selection of creative design concepts using formal and informal methodologies; Development of simple mathematical models for final design concept; Utilization of rough prototyping methods; Consideration of safety, ethical, and environmental issues; Understanding when to select off-the-shelf components versus when to fabricate custom components; Understanding the relationship between design and manufacturing, incuding the selection of appropriate manufacturing processes Use of physical and/or virtual prototypes of sufficient detail to serve the purpose of proof-of-concept and recommendations for design improvement Presentation and reporting of final project outcomes and recommendations
COURSE STRUCTURE/SCHEDULE: Lecture: 2 days per week at 1.5 hours, Laboratory: 2 days per week at 3 hours	oratory: 2 days per week at 3 hours

	 Regular written and/or oral design reviews Project exposition Peer review 	ASSESSMENT TOOLS: for each assessment tool, links to the course outcomes are identified
	 Given a qualitative and open-ended "real-world" engineering design problem, suggest a solution based on technical analysis Learn to work effectively in engineering teams to resolve conflict and meet quantitative engineering objectives established during the project. Learn to communicate effectively with peers, project sponsors, advisors, and/or mentors Learn to consider unstructured creativity as a natural part of a structured design process and to systematically generate concepts using methods such as brainstorming and decomposition Learn to make appropriate assumptions and exercise engineering judgment in solving an open-ended problem Manage and plan large design projects using time management tools, and be able to handle uncertain and incomplete information effectively to meet project goals Learn to clearly request and exchange quantitative information, and to communicate project results, to audiences of varying expertise levels Learn patent and literature search methods, benchmarking, and other general forms of background independent learning Integrate past course material to advance basic system concepts to a prototyping level, providing support for all design decisions by defensible engineering analysis and reasoning 	COURSE OUTCOMES: for each course outcome, links to the Course Objectives are identified in brackets.
1	 Solve an open-ended mechanical engineering design problem including considerations of performance, cost, and societal considerations. The problem must provide opportunities for creative mechanical design, fundamental analysis, and proof-of-concept prototyping. Each student team works on a different project and everyone participates in project proposal development, reporting, and final design presentations [3, 4, 5, 6, 7, 8, 10] Apply a design process appropriate to the engineering problem at hand, including unstructured creativity as part of a structured design problem [3, 5] Generate and evaluate design concepts after gaining a sound understanding of the problem background and existing design concepts [1, 3, 5] Identify a set of design variables and governing equations for the selected design concept that can be utilized to improve the design [3, 5, 11] 	COURSE OBJECTIVES: for each course objective, links to the Program Outcomes are identified in brackets.

THE UNIVERSITY OF MICHIGAN -- COLLEGE OF ENGINEERING Course Approval Request

College Curriculum Committee, 1420 Lurie Engineering Center Building

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Form	Number
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Action Requested

New CourseModification of Existing CourseDeletion of Course

Complete the following sections:
New Courses - B & C completely
Modifications - A modified information, B & C completely
Deletions - A & C completely

Date 6/29/2006 Effective Fall 2006

	A. CU	RRENT LISTIN	G			B. RI	EQUESTED LIST	ING		
	Home Depart	tment		Div#	Course Number	Home Dep	artment		Div #	Course Number
						Mechanic	cal Engineering			495
	Cross Listed 0	Course Information				Cross Listed	Course Information			
	Course Title			***************************************		Course Title)			
	TITLE ABBRE- VIATION	Time Sched Max = 19 Spaces Transcript Max = 20 Spaces				TITLE ABBRE- VIATION	Time Sched Max = 19 Spaces Transcript Max = 20 Spaces	LAB II		
	Course Descr					Weekly to demo applied controls mechan laborate	ription for Official Publication lectures and extenstrate experime to complex medias, heat transfer, lics, materials, arony report writing skills, and the discontinuous and the discontinuous control of the discontinuous	ended experime lental and analy hanical systems fluid mechanics nd dynamical sy g, oral presenta	tical meth . Topics v , thermody stems. En tions, and	ods as vill include vnamics, nphasis on
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Х	Prerequisites	ME360. ME395: pre ⊙ Enforced ○ Advise	eceded or accompanied	bv ME350			ME360, ME395, P/A M © Enforced Advis	ed		
Х	Credit Restrictions	Recommend ME495 no	ot be elected concurre	ntly. Not open	to graduate	Credit Restrictions	May not elect ME450 c	oncurrently. Not open	to graduate st	udents.
	Level of Cree Undergrad Rackham Non-Rckh	dit donly ☐ Ugrad of Grad ☐ All Creo m Grad ☐ Rokhm	or Non-Rckhm Grad dit types Grad w/add'l Work	Credit Hours Min Max	Contact Hrs/Wk Number of Wks	Level of Cr ☐ Undergra ☐ Rackham ☐ Non-Rckl ☐ Ugrad or	d only Ugrad or All Credit	Non-Rckhm Grad types Grad w/add'l Work	Credit Hours Min Max 4 4	Contact Hrs/Wk 5 Number of Wks 14
C.	Is this cours Maximum	se repeatable? O Yes	aximum Times?	-	•	Printing	Information Print the (Optional) Print the		dule	
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Adding preceded or accompanied by ME335 is necessary because ME495 includes one or more heat transfer based laboratories. Consequently, students need to have a working knowledge of heat transfer for these labs.
The ME program does not recommend that students take ME450 and ME495 concurrently. For the past couple of years, this has been communicated to students on an individual basis when they come to ASO to talk about their studies. For students who want to ake both, they are asked to seek the approval of the UG Program Director by discussing their situation, why they want/need to do so typically, the course load is evaluated so that it, at most, includes ME450, ME495 and no more than one more technical course.
At issue is that the ME450/ME495 combination not only adversely impacts the particular student taking these courses, but even more importantly, can adversely impact the group members in each of these courses.
The proposed change would formalize the existing advising process and better help students develop their plan of study.
Are any special resources or facilities required for this course?
Detail the Special requirements
Unchanged since 1998-99

ME495 COURSE PROFILE DEGREE PROGRAM: Mechanical Engineering

COURSE NUMBER: ME495	COURSE TITLE: Laboratory II
REQUIRED COURSE OR ELECTIVE COURSE: Required	TERMS OFFERED: Fall, Winter
TEXTBOOK / REQUIRED MATERIAL:	PRE / CO-REQUISITES: ME 395: Laboratory I; ME 360: Modeling, Analysis, and Control of Dynamic Systems; preceded or accompanied by ME 335. May not elect concurrently with ME 450: Design and Manufacturing III
COGNIZANT FACULTY: A. M. Sastry	COURSE TOPICS:
BULLETIN DESCRIPTION: Weekly lectures and extended experimental projects designed to demonstrate experimental and analytical methods as applied to complex mechanical systems. Topics will include controls, heat transfer, fluid mechanics, thermodynamics, mechanics, materials, and dynamical systems. Emphasis on laboratory report writing, oral presentations, and team-building skills, and the design of experiments.	 Frequency Response Analysis of a Flexible Torsional System Choose either a or b: (a) Evaluation of a Vapor/Compression Cycle; (b) Dynamic Compensation of a Subwoofer Speaker System Choose either a, b, or c: (a) Performance of a Single Cylinder Engine; (b) Injection Molding of Plastic Parts; (c) Evaluation of the Welding Process Student Designed Laboratory
COURSE STRUCTURE/SCHEDULE: Lecture: 2 days per week at 1.5 hours, Laboratory: 1 day per week at 3.0 hours	oratory: 1 day per week at 3.0 hours

COURSE OUTCOMES: 1. Investigate how to compare the mathematical model of a dynamic system with its experimental performance [1, 2] for active course and evaporation, and cooling capacity of a vapor compression cycle, evaluate the effectiveness of limits to the compressor over a range of operating conditions [1, 2]. b. Experimentally measure the frequency design a feedbackfeed forward compensator, implement the compensator will an advanced as subvooler speaker, obtain a simplified model of the speaker, modern through compensator will an advanced as a simplified model of the speaker, modern the compensator will appear the compensator will be a simplified model of the speaker, modern the compensator implement the compensator will and mechanical strong the speaker for airfule, load, and compression ratios on engine performance and emissions. Evaluate the effect of airfule, load, and compression ratios on engine performance and emissions. Evaluate the engine performance of an injection moding process. Examine the dimensional streams and flow patterns in clear parts using optian intended. Examine the mechanical streams and flow patterns in clear parts using optical methods. Examine the world consist of the mechanical streams and flow patterns in clear parts using optical methods. Examine the world consist of an experimental program [3]. 4. Create a proposal for an experimental program [3]. S. Plan and execute an experimental program [3]. S. Plan and a secure and experimental program [3]. S. Plan and a secure and experimental program [3]. S. Plan and a secure and experimental program [4]. Self-experimental program [5]. S. Plan and

PREPARED BY: W. Schultz LAST UPDATED: September 22, 2006

THE UNIVERSITY OF MICHIGAN -- COLLEGE OF ENGINEERING Course Approval Request

College Curriculum Committee, 1420 Lurie Engineering Center Building



Form	Number
1749	

Action Requested

New CourseModification of Existing CourseDeletion of Course

Complete the following sections:
New Courses - B & C completely
Modifications - A modified information, B & C completely
Deletions - A & C completely

Date 6/29/2006 Effective Winter 2007

	A. CI	JRRENT LISTIN	IG			B. R	EQUESTED LIST	TING		
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This new course is in response to instructional need for a unified, atomic understanding of heat transfer and its modern, quantum- nechanics based applications. The course objection, key concepts, and syllabus are attached. The course was taught in Winter 2005 and Winter 2006 with enthusiastic response (Q1=4.0, Q2=4.0) and is planned for Winter 2007.	
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Are any special resources or facilities required for this course?	•
Detail the Special requirements	



ME 599-001, **HEAT TRANSFER PHYSICS**, Winter 2007 Tu & Th, 11:30-1:00, 1371 G.G. Brown

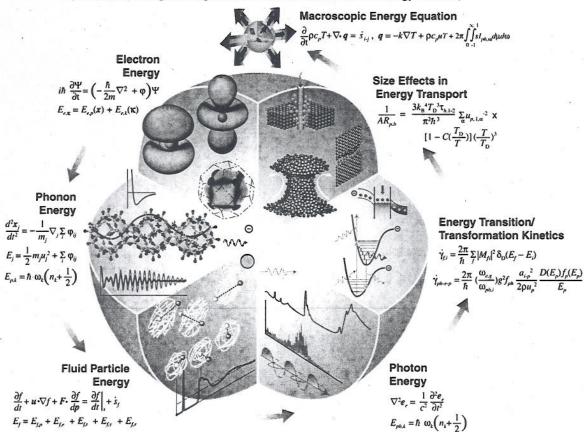
3 Credits, Prerequisite: Heat Transfer (ME 335 or equivalent), Instructor: M. Kaviany Department of Mechanical Engineering, University of Michigan, Ann Arbor

OBJECTIVE

Heat transfer physics describes the kinetics of storage, transport, and transformation of microscale energy carriers (phonon, electron, fluid particle, and photon). Sensible heat is stored in the thermal motion of atoms, in various phases of matter. The atomic energy states and their populations are described by the classical and the quantum statistical mechanics (partition function) and the combinatoric energy distribution probabilities. Transport of thermal energy by the microscale carriers is based on their particle, quasi-particle, and wave descriptions, their diffusion, flow and propagation, and their scattering and transformation they encounter as they travel. The mechanisms of energy transitions amongst these energy carriers, and their rates (kinetics), are governed by the match of their energies, their interaction probabilities, and the various hindering-mechanism rate limits. Conservation of energy describes the interplay amongst energy storage, transport, and conversion, from the atomic to the continuum scales. The figure below renders the fundamentals of heat transfer physics.

Heat Transfer Physics

(Kinetics of Storage, Transport, and Transition of Microscale Energy Carriers)



The course combines the fundamentals (through survey summaries) of interatomic potential, statistical thermodynamics, computational molecular dynamics (include lattice dynamics, with computer codes) quantum mechanics, transport theories (including Boltzmann and stochastic transport), solid-state physics (including semiconductors), and radiation transport (including spontaneous and stimulated emissions), as related to heat transfer and thermal energy conversion. The aim is to present a unified theory of heat transfer physics and its modern applications (through case studies). The course is suitable for students in engineering, applied physics, and physical chemistry. The prerequisite is undergraduate education in engineering or science. The outline of the course is given below.

OUTLINE

1 Introduction and Preliminaries

- 1.1 Phonon, Electron, Fluid Particle, and Photon
- 1.2 Combinatorial Probabilities and Energy Probability Distribution Function
- 1.3 Particle, Wave, Wave Packet, and Quasi-Particle
- 1.4 A History of Contributions towards Heat Transfer Physics
- 1.5 Energy in Classical and Quantum Mechanics
- 1.6 Periodic Table of Elements
- 1.7 Heat Transfer Physics
- 1.8 Scope

2 Interatomic Potentials, Molecular Dynamics, and Quantum Energy States

- 2.1 Interatomic Forces and Potential Well
- 2.2 Interatomic Potential Models
- 2.3 Molecular Ensembles, Temperature, and Thermodynamic Relations
- 2.4 Classical Mechanics, Hamiltonian, and Partition Function
- 2.5 Molecular Dynamics Simulations
- 2.6 Schrödinger Equation and Quantum Mechanics

3 Carrier Energy Transport and Transformation Theories

- 3.1 Boltzmann Transport Equation
- 3.2 Energy Transformation Kinetics and Fermi Golden Rule
- 3.3 Maxwell Electromagnetic Wave Propagation Equations
- 3.4 Onsager Transport Coefficients
- 3.5 Stochastic Particle Dynamics and Transport
- 3.6 Green-Kubo Transport Theory
- 3.7 Macroscopic Fluid Dynamics Equations
- 3.8 Macroscopic, Elastic Solid Mechanics

4 Phonon Energy Storage, Transport and Transformation Kinetics

- 4.1 Phonon Dispersion in One-Dimensional, Classical Lattice Vibration
- 4.2 Phonon Density of States
- 4.3 Reciprocal Lattice and Brillouin Zone
- 4.4 Three-Dimensional Lattice Dynamics and Dispersion Relation
- 4.5 Quantum Theory of Lattice Vibration
- 4.6 Examples of Phonon Dispersion and DOS
- 4.7 Debye Lattice Specific Heat Capacity
- 4.8 Atomic Displacement in Lattice Vibration
- 4.9 Phonon BTE and Callaway Conductivity Model
- 4.10 Einstein and Cahill-Pohl Minimum Conductivities
- 4.11 Phonon Conductivity from MD and G-K Autocorrelation
- 4.12 Quantum Corrections to MD Predictions
- 4.13 Phonon Conductivity from BTE: Variational Method
- 4.14 Optical Phonon Contribution to Conductivity
- 4.15 Experimental Data on Phonon Conductivity
- 4.16 Phonon Boundary Resistance
- 4.17 Absorption of Ultrasound Waves in Solids

5 Electron Energy Storage, Transport and Transformation Kinetics

- 5.1 Electron Band Structure in Crystals
- 5.2 Electronic Energy Bands in One-Dimensional Ionic Lattice
- 5.3 Three-Dimensional Bands Tight-Band Approximation
- 5.4 Electron Band Structures for Semiconductors and Effective Mass
- 5.5 Fermi Energy
- 5.6 Drude-Sommerfeld Electron Gas Model for Metals
- 5.7 Density of Electronic Energy States for Semiconductors
- 5.8 Electron Specific Heat Capacity
- 5.9 Electron BTE for Semiconductors
- 5.10 Scattering Potential, Fermi Golden Rule, and Relaxation Time
- 5.11 Average Relaxation Time for Power-Law Energy Dependent Relaxation Time
- 5.12 Transport Coefficients in Coupled Electrical and Thermal Currents
- 5.13 Semiconductor Electro-Thermal Transport Properties using j_e and Gradient of T
- 5.14 Magnetic Field and Hall Factor
- 5.15 Phonon Scattering of Electron
- 5.16 Electro-Thermal Transport Properties Data for Semiconductors and Metals
- 5.17 Electron and Phonon Transport Equations under Thermal Non-Equilibrium
- 5.18 Cooling Length in Electron- Lattice Thermal Non-equilibrium
- 5.19 Electronic Energy States of Rare-Earth Ion Doped Crystals
- 5.20 Electronic Energy States of Gases

6 Fluid Particle Energy Storage, Transport and Transformation Kinetics

- 6.1 Fluid Particle Quantum Energy States and Partition Functions
- 6.2 Idea-Gas Specific Heat Capacity
- 6.3 Dense-Fluid Specific Heat Capacity
- 6.4 Fluid Particle BTE and Average Molecular Speed
- 6.5 Elastic Binary Collision Rate of Ideal Gas
- 6.6 Ideal Gas Mean-Free Path
- 6.7 Relaxation Time Approximation of Gas BTE
- 6.8 Thermal Conductivity of Ideal Gas
- 6.9 Thermal Conductivity of Liquids
- 6.10 Effective Conductivity with Suspended Particles in Brownian Motion
- 6.11 Interaction of Fluid Particle and Surface
- 6.12 Turbulent-Flow Structure and Transport
- 6.13 Thermal Plasmas

7 Photon Energy Storage, Transport and Transformation Kinetics

- 7.1 Quasi-Particle Treatment: Photon Gas and Planck Emission
- 7.2 Laser and Narrow-Band Emissions
- 7.3 Classical and Semi-Classical Treatments of Photon-Matter Interaction
- 7.4 Photon Absorption and Emission in Two-Level Electron Energy System
- 7.5 Photon BTE: Scattering, Absorption, and Emission
- 7.6 Equation of Radiative Transfer
- 7.7 Continuous and Band Photon Absorption in Solids
- 7.8 Continuous and Band Photon Emission in Solids
- 7.9 Spectral Surface Emission
- 7.10 Radiative and Non-Radiative (with Phonon Emission) Decays and Quantum Efficiency
- 7.11 Anti-Stokes Fluorescence and Photon-Electron-Phonon Couplings
- 7.12 Role of Phonon in Photovoltaic
- 7.13 Role of Fluid-Particle Motion in Gas-Phase Lasers and Laser Cooling of Gases

Course Materials: Course Pack: Heat Transfer Physics, 2006, is provided (pdf).

Key Concepts in Heat Transfer Physics (ME 599-001)

Atomic Description of Energy and Its Transport and Transformation

- Four microscale carriers: phonon, electron (hole), fluid particle, and photon
- · Particle, waves, ,wave packets, and quasi-particle behaviors: quantum mechanics
- Interatomic potential (Schrödinger equation-ab initio calculations) and its models
- Molecular dynamics simulations and scales in molecular motion
- Schrödinger equation and quantum harmonic oscillator and electron gas
- Boltzmann transport equation (BTE), particle scattering (interaction)
- · Scattering (interaction) rate kinetics and Fermi golden rule
- Onsager coupled-transport theorem
- Stochastic particle dynamics and transport (Langevin equation)
- Green-Kubo transport coefficients (fluctuation-dissipation correlations)
- Maxwell equations (including EM wave equation and relation to photon)
- Macroscopic conservation equations (energy, fluid dynamics, and elastic solid mechanics)

Phonon Energy Storage, Transport and Transformation Kinetics

- Lattice and its vibration (phonon dispersion, band gap, and density of states)
- Phonon heat capacity (Debye model)
- Phonon BTE and thermal conductivity (Callaway model)
- · Phonon scattering mechanisms and relaxation time models
- Einstein and Cahill-Pohl minimum thermal conductivity models
- Phonon conductivity from MD and Green-Kubo autocorrelation
- Phonon conductivity from variational principles
- · Phonon boundary resistance (diffuse and specular) using photon treatment

Electron Energy Storage, Transport and Transformation Kinetics

- Electrons in solids, allowed states, band gaps and band-gap models
- Electron density of states and heat capacity
- Electron BTE and electro-thermal transport properties
- · Electron-phonon interaction rate and the Fermi golden rule
- · Other mechanisms of electron scattering and their relaxation time models
- Electronic thermal conductivity of solids (Wiedemann-Franz-Lorenz relation)
- · Thermoelectricity (Seebeck and Peltier coefficients) and its figure of merit
- Electron-phonon thermal nonequilibrium and cooling length

Fluid Particle Energy Storage, Transport and Transformation Kinetics

- Gas and liquid heat capacity (energy partition function)
- · Gas particle BTE, collision rate, relaxation time, and mean-free path
- Gas thermal conductivity from BTE
- · Liquid thermal conductivity (random, localized motion)
- Conductivity of suspended particles in liquid (Brownian motion and nanofluid conductivity)
- Gas molecule-surface interaction (fluid flow regimes and accommodations and slips)
- Solid particle thermophoresis in gases
- Physical surface adsorption and desorption of gas molecules
- Turbulent-flow structure and transport
- Thermal plasmas

Photon Energy Storage, Transport and Transformation Kinetics

- · Harmonic oscillator and blackbody photon emission intensity
- Laser emission
- Photon absorption and emission (spontaneous and stimulated) in two-level electronic systems
- Photon BTE (photon and phonon Equation of Radiative Transfer)
- Mechanisms of spectral absorption in solids (including semiconductors and metals)
- Mechanisms of spectral absorption in gases
- · Radiative and non-radiative (with phonon emission) decays and quantum efficiency
- Photon-electron-phonon couplings and laser cooling of solids
- · Role of Phonon in photovoltaic and extraction of hot electrons
- Role of Fluid-particle kinetic energy in gas laser (including laser cooling of atomic gases)

ME 599-001 (Heat Transfer Physics), Tu & Th 11:30-1:00, Winter 2007, 1371 G.G. Brown Department of Mechanical Engineering, University of Michigan

WEEK	SUBJECT	READING (Chapter. Section)	HOMEWORK (Assigned)
January 1	Introduction; Macroscopic Energy Equation and Role of Microscale (Atomic-Level) Heat Carriers: Electron, Phonon, Fluid Particle, and Photon; Atomic-level Energy Kinetics: Length, Time, and Energy Scales; Scope of Heat Transfer Physics	1.1 - 1.8	HW # 1
January 8	Interatomic Potentials, Molecular Dynamics, and Quantum Energy States; Interatomic Forces, Potentials and Models; Ab Initio Interatomic Potential Calculations and Models; Statistical Ensembles, Energies, Temperature, and Partition Function; Hamiltonian Mechanics	2.1 - 2.4	HW # 2 Interatomic Potential Gaussian Code
January 15	Computational, Classical Molecular Dynamics Simulation; Schrödinger Equation; Quantum Simple Harmonic Oscillator and for Free Electron Gas	2.5 - 2.6	HW # 3 2-D MD Gas Particle Code
January 22	Carrier Transport and Interaction Theories; Boltzmann Transport Equation (BTE); In- and Out-Scattering; Relaxation Time Approximation; Scattering Rate Approximation, Energy Interaction Rates and Fermi Golden Rule	3.1 - 3.2	HW # 4
January 29	Maxwell Equations; Onsager Transport Coefficients; Stochastic Transport Processes (Langevin Equation); Green-Kubo (G-K) Autocorrelation Decay and Thermal Conductivity; Macroscopic Fluid Dynamics Equations; Elastic Solid Mechanics Equations	3.3 - 3.8	HW # 5 3-D MD-GK Code for Ar FCC Phonon k
February 5	Phonon; Dispersion in Harmonic Lattice Vibration (Acoustic and Optical Phonons); Phonon Density of States; Reciprocal Lattice Space and Brillouin Zones; Dynamical Matrix SELECTION OF PROJECT TOPIC	4.1 - 4.6	HW # 6 3-D Ar FCC Phonon Dispersion Code
February 12	Lattice Specific Heat Capacity; Phonon BTE and Callaway Conductivity Model (Single Mode Relaxation Time); MD-GK (Heat Current Autocorrelation); Cahill-Pohl Minimum Conductivity, Variational Method; Phonon Boundary Resistance	4.7 - 4.16	HW # 7 Case Study I
February 19	Electron; Band Structure in Crystals; Tight-Binding Approximation; Full and Model Band Structures and Effective Electron Mass in Semiconductors; Fermi Energy in Electron Gas Model, Electron Density of States; Drude-Sommerfeld Model for Metals; Electron Specific Heat Capacity ABSTRACT OF PROJECT	5.1 - 5.8	HW # 8 Fermi Surface Code 3-D SiC Electronic Band Structure Code

February 26

SPRING RECESS

March 5	Electron BTE for Semiconductors; Energy-Dependent Relaxation Time; Semiconductor and Metal Electro-Thermal Transport Coefficients (Electrical Conductivity, Seebeck Coefficient, Peltier Coefficient, and Electronic Thermal Conductivity); Electron	5.9 - 5.22	HW # 9
	Scattering by Phonon; Electron-Lattice Thermal Non-Equilibrium and Cooling Length		Case Study II
March 12	Fluid Particle; Quantum Fluid Particle Electronic, Translational, Vibrational, and Rotational Energy States and Partition Functions; Fluid Particle Specific Heat Capacity (Ideal Gas and Dense Fluids); Fluid Particle BTE; Equilibrium Energy Distribution Function; Binary Collision Rate; Relaxation Time and Mean-Free Path	6.1 - 6.7	HW # 10 3-D Liquid MD Code for Fluid Particle Trajectory
March 19	Ideal Gas Thermal Conductivity from BTE; Liquid Thermal Conductivity; Brownian Motion of Solid Particles and Effective Conductivity; Fluid Particle-Surface Interaction and Flow Regimes; Thermophoresis; Adsorption and Desorption; Turbulent- Flow Structure and Transport; Thermal Plasmas	6.8 - 6.13	HW # 11 3-D Surface Accommodation Coefficient MD Code
March 26	Photon; Planck Distribution for Photon Gas (Blackbody Radiation); Lasers and Narrow Band Emission; Photon Absorption and Stimulated and Spontaneous Emissions in Two-Level Electronic Transition System; Electronic Population Rage Equation; Photon BTE and Equation of Radiative Transfer Using Absorption, Emission, and Scattering Cross Sections; Radiant Thermal Conductivity	7.1 - 7.6	HW # 12
	DRAFT REPORT ON PROJECT		
April 2	Spectral Continuous and Band Photon Absorption and Emission in Solids; Absorption Coefficient for Metals and Semiconductors; Relation between Absorption and Emission in Multi-Level Electron System	7.7 - 7.9	HW # 13
April 9	Radiative and Non-Radiative (Involving Phonon Emission) Decays and Quantum Efficiency; Anti-Stokes Fluorescence and Photon- Electron-Phonon Couplings (Laser Cooling of Solids); Role of Phonon in Photovoltaic; Role of Fluid Particle Motion in Gas Lasers (Laser Cooling of Gases)	7.10 - 7.13	Case Study III
April 16	Review, Course Evaluation Last Day of Class: Tuesday, April 17		
April 23	FINAL REPORT ON PROJECT DUE TUESDAY APRIL 24		
Instructor:	Massoud Kaviany, Office: 3108 G.G. Brown, Phone: 936-0402, E-mail: kaviany@umich.edu		
Office Hours:	MW: 12:30 - 1:30, and by Appointment		
Grade Policy:	Homework: 70% Project: 30% (Projects will be graded based on heat transfer physics content)		
Course Pack:	Heat Transfer Physics , Course Pack, M. Kaviany, 2006		

THE UNIVERSITY OF MICHIGAN -- COLLEGE OF ENGINEERING Course Approval Request

College Curriculum Committee, 1420 Lurie Engineering Center Building



Form	Numbe
1772	

Action Requested

New Course
 Modification of Existing Course
 Deletion of Course

Complete the following sections:
New Courses - B & C completely
Modifications - A modified information, B & C completely
Deletions - A & C completely

Date 10/10/2006 Effective Winter 2007

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SUPPORTING STATEMENT

This course prepares the engineering student to meet the ever-increasing educational demands of the rapidly growing nanotechnology field. The application of nanotechnology may revolutionize many engineering fields and our society. However, it relies
popully on the education of future engineers. This course is designed for graduate students from various engineering programs. It is
self contained. The concents, way of thinking, and methods of attacking problems are applicable to a wide range of disciplines. Due
o its interdisciplinary nature, the course has integrated phenomena and processes in several fields, including solid mechanics, materials, physics and electronics. The focus of the course is scientific understanding and computational techniques. The key element
of my nanotechnology education is vigorous participation of students in class discussion of the subjects at hand.
A group project is an important component of this course. The studentis encouraged to choose any nano-related topic relevant to the research fields of the group members, as long as the topic can be addressed with the approaches taught in the class. These projects have demonstrated high quality, and contributed to students' research work. This course has provided a unique opportunity for students with different backgrounds to work together.
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The course was first taught in W02 and has been taught four (4) times, which enthusiastic response (Average Q1=4.51 and Q2=4.49 and Highest Q1=4.88 and Q2=4.88) and is planned for W07. The course syllabus is attached. For previous projects, please check website: http://www-personal.umich.edu/~weilu/me599/
Senior undergraduate students may also join the class by instructor permission.
Are any special resources or facilities required for this course? ☐ Yes ☒ No
Detail the Special requirements

THE UNIVERSITY OF MICHIGAN -- COLLEGE OF ENGINEERING Course Approval Request

College Curriculum Committee, 1420 Lurie Engineering Center Building



Form	Number
1652	

Action Requested

New Course
 Modification of Existing Course
 Deletion of Course

Complete the following sections:
New Courses - B & C completely
Modifications - A modified information, B & C completely
Deletions - A & C completely

Date 9/20/2006
Effective Winter 2007

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brough the development and use of physically-based mathematical mo-	dels and associated measurement technique	s for model
verification and process monitoring. After completing the course, studen	nts should be able to determine whether or n	ot a model is
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nocess parameters, and develop their own processes. These skills are	they find the knowledge gained to be not only	u ucoful for its
n Materials Processing. Course feedback from students indicates that i	they find the knowledge gained to be not only	y userui lor its
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Are any special resources or facilities required for this course?		
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